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(54) **Bleaching compositions**

(57) The present invention relates to a liquid bleaching composition comprising a peroxy-carboxylic acid

having at least one aromatic group and a co-agent selected from organic acids, homo or copolymers of acidic monomers and mixtures thereof.

**EP 1 010 751 A2**

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**Description**Technical field

- 5 **[0001]** The present invention relates to a bleaching compositions which can be used to bleach various surfaces including but not limited to, hard-surfaces as well as fabrics, clothes, carpets and the like.

Background of the invention

- 10 **[0002]** Commonly encountered liquid aqueous cleaning compositions for removal or cleaning stains on fabrics and/or hard-surfaces may comprise a bleaching agent. A number of bleaching agents are known in the art. Halogen bleaches whilst being extremely effective bleaching agents, often also present a number of drawbacks which can sometimes dissuade a consumer from choosing the halogen-containing product. For example halogen bleaches, especially chlorine bleaches, emit a pungent odour during and after use (e.g., on consumer hands and/or surfaces treated therewith)  
15 which some consumer find disagreeable.

- [0003]** Furthermore, it is known in the art that halogen bleach-containing compositions (typically hypochlorite) are relatively aggressive to fabrics and may cause damage when used in relatively high concentration and/or repeated usage. In particular the consumer may perceive damage to the fabric itself (e.g. loss of tensile strength) or damage to the colour intensity of the fabric. While colour and fabric damage may be minimised by employing milder oxygen bleaches such as hydrogen peroxide or hydrogen peroxide generating bleaches, the bleach performance characteristics of such peroxygen bleaches are much less desirable than those of the halogen-containing bleaching agents. Therefore, liquid aqueous activated peroxygen bleach-containing compositions have been developed containing activators, i.e., compounds which enhance peroxygen bleaching performance. However these bleaches do not perform as well as hypochlorite bleaches in stain removal.

- 20 **[0004]** Peroxycarboxylic acid (peracid) bleaches are known in the art. The Applicant has found that such bleaches, in this instance, are particularly useful as they do not present the disadvantages of hypochlorite bleaches and are better performing bleaching agents than the oxygen bleaches. However generally peracid bleach-containing compositions have stability problems as the peracid prematurely begins to degrade. This premature degradation of the peracid is believed to be due to the presence of trace amounts of metal ions in the formulation which may catalyse the further  
25 degradation of the peracid.

**[0005]** The present invention thus provides a bleaching composition which is stable, safe to the surfaces treated and delivers improved bleaching performance when used in laundry applications and/or in any household application (e.g. bleaching/disinfecting of hard-surfaces).

- 30 **[0006]** Furthermore, it is believed that the solubility of the particulate pre-formed peracid is affected by the level of hardness ions. By 'hardness ions' it is meant ions that are present in water and form insoluble salts that precipitate from solution. The most predominant hardness ions are calcium and magnesium ions. The water may be either water used to prepare the composition of the present invention or may be the wash water into which the composition of the present invention is dispensed. Thus, it is another object of the present invention to prevent or at least reduce the precipitation of the peracid from the composition.

- 40 **[0007]** The compositions according to the present invention may be useful in any laundry application, e.g., as a laundry detergent or a laundry additive, and when used as a laundry pre-treater. A particular advantage of the compositions of the present invention is that they are suitable for the bleaching of different types of fabrics including natural fabrics, (e.g., fabrics made of cotton, and linen), synthetic fabrics such as those made of polymeric fibres of synthetic origin (e.g., polyamide-elasthane) as well as those made of both natural and synthetic fibres. For example, the bleaching  
45 compositions of the present invention herein may be used on synthetic fabrics despite a standing prejudice against using bleaches on synthetic fabrics, as evidenced by warnings on labels of clothes and commercially available bleaching compositions like hypochlorite-containing compositions.

- [0008]** Another advantage of the bleaching compositions according to the present invention is that they can be used in a variety of conditions, i.e., in hard as well as soft water and in either neat or diluted form. More particularly, it has been found that the liquid aqueous compositions of the present invention find a preferred application when used in their diluted form in any application and especially in any conventional laundry application. Indeed, upon dilution (typically at a dilution level of 20ml/L or more (composition:water)) the compositions of the present invention become less acidic, e.g., from a pH of about 1.5 to about 6.5 or more. The compositions according to the present invention although delivering effective bleaching performance in their neat form surprisingly exhibit further enhanced bleaching performance  
50 in their diluted form. Actually, this "pH jump" effect allows to formulate acidic liquid aqueous compositions (i.e. pH below 7, preferably below 3 and more preferably below 2) which are physically and chemically stable upon prolonged periods of storage and which deliver outstanding bleaching performance under diluted usage conditions.

**[0009]** Yet another advantage of the compositions of the present invention is that they exhibit also effective stain

removal performance on various stains including enzymatic stains and/or greasy stains.

### Summary of the invention

[0010] According to the present invention there is provided a liquid detergent composition comprising a pre-formed particulate peroxy carboxylic acid containing at least one aromatic group and a co-agent selected from the group consisting of organic acid, homo and copolymer of acidic monomers and mixtures thereof.

[0011] In another preferred aspect of the present invention there is provided a liquid bleaching product comprising a first composition comprising a peroxy carboxylic acid containing at least one aromatic group and a second composition comprising a surfactant wherein the product is packaged such that first and second compositions are separated from each other.

[0012] The present invention further encompasses methods of bleaching a surface with a composition according to the present invention.

### Detailed description of the invention

#### Liquid bleaching composition

[0013] The compositions according to the present invention are liquid and preferably aqueous. Thus the composition preferably has a water content of from 10% to 99%, more preferably from 30 to 95%, more preferably 40 to 90% by weight of the composition.

[0014] Where the composition according to the present invention is liquid is preferably has a pH of below 7. Preferably, the pH of the compositions according to the present invention is from 1 to 6, more preferably from 1 to 5, even more preferably from 2 to 4. Formulating the compositions according to the present invention in the acidic pH range is critical to the chemical stability of the compositions according to the present invention. The pH of the composition is preferably below the pKa of the peracid used. It is believed that the acidic pH controls/limits the formation of highly reactive species which are unstable in acidic medium upon storage, and thus contributes to the stability of the compositions for prolonged periods of storage.

[0015] The pH of the compositions may be adjusted by any acid or alkaline species known to those skilled in the art. Examples of acidic species suitable for use herein are organic acids, such as citric acid and inorganic acids, such as sulphuric acid, sulphononic acid and/or metanesulphonic acid. Examples of alkaline species are sodium hydroxide, potassium hydroxide and/or sodium carbonate.

[0016] It has been found that the bleaching performance of the composition of the present invention is significantly improved versus the same composition but without either the peroxy carboxylic acid and/or the co-agent. In fact it has been found that a synergy has been identified that could not have been foreseen from bleaching performance data of compositions comprising either the peroxy carboxylic acid or the co-agent alone.

[0017] The bleaching performance of the present composition may be evaluated by the following test methods on various type of bleachable stains.

[0018] A suitable test method for evaluating the bleaching performance on a soiled fabric under diluted conditions is the following: A composition according to the present invention is diluted with water typically at a dilution level of 1 to 100 ml/L, preferably 20 ml/L (composition :water), then the soiled fabrics are soaked in it for 20 minutes to 6 hours and then rinsed. Alternatively the bleaching composition can be used in a washing machine at a dilution level of typically at a dilution level of 1 to 100 ml/L (composition :water). In the washing machine the soiled fabrics are washed at a temperature of from 30° to 70°C for 10 to 100 minutes and then rinsed. The reference composition in this comparative test undergoes the same treatment. Soiled fabrics/swatches with for example tea, coffee and the like may be commercially available from E.M.C. Co. Inc..

[0019] The bleaching performance is then evaluated by comparing side by side the soiled fabrics treated with a composition of the present invention with those treated with the reference, e.g., the same composition but comprising no bleach or a different bleach. A visual grading may be used to assign difference in panel units (psu) in a range from 0 to 4.

[0020] An advantage of the compositions of the present invention is that they are physically and chemically stable upon prolonged periods of storage.

[0021] Chemical stability of the compositions herein may be evaluated by measuring the concentration of available oxygen at given storage time after having manufactured the compositions. By "chemically stable", it is meant herein that the compositions of the present invention comprising a peracid do not undergo more than 15% AvO loss, in one month at 25°C and preferably not more than 10%.

[0022] The loss of available oxygen (AvO) of a peracid-containing composition over time can be measured by titration with potassium permanganate after reduction with a solution containing ammonium ferrous sulphate. This stability test

method is well known in the art and is reported, for example, on the technical information sheet of Curox<sup>R</sup> commercially available from Interlox. Alternatively peracid concentration can also be measured using a chromatography method described in the literature for peracids (F. Di Furia et al., Gas-liquid Chromatography Method for Determination of Peracids, Analyst, Vol 113, May 1988, p 793-795).

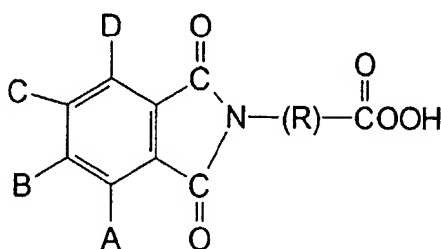
[0023] By "physically stable", it is meant herein that no phase separation occurs in the compositions according to the present invention for a period of 7 days at 50°C.

#### Peroxy carboxylic acid

[0024] The bleaching composition of the present invention comprises a pre-formed particulate peroxydicarboxylic acid containing at least one aromatic group (hereinafter referred to as peracid).

[0025] Preferred peracids are those having general formula  $A_m-Q-A$  wherein Q is an aromatic group; A is  $P-(R)_n-C(O)OOH$  wherein R is C1-10 alkyl, preferably C1-4 alkyl; n is from 1 to 8, preferably 1 to 5; P is either CO-NH or C(O)NC(O); and m is either 0 or 1.

[0026] In a preferred peracid are those having general formula:

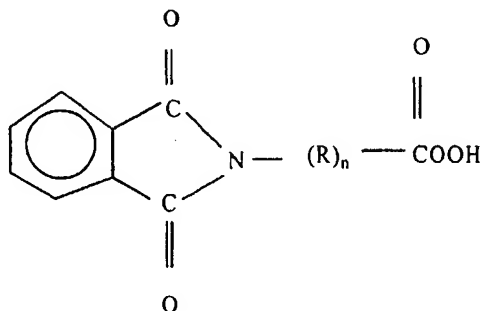


where R is C1-20 alkyl group and where A, B, C and D are independently either hydrogen or substituent groups individually selected from the group consisting of alkyl, hydroxyl, nitro, halogen, amine, ammonium, cyanide, carboxylic, sulphate, sulphonate, aldehydes or mixtures thereof.

[0027] In a preferred aspect of the present invention R is an alkyl group having from 3 to 12 carbon atoms, more preferably from 5 to 9 carbon atoms. Preferred substituent groups A, B, C and D are linear or branched alkyl groups having from 1 to 5 carbon atoms, but more preferably hydrogen.

[0028] Preferred peracids are selected from the group consisting of phthaloyl amido peroxy hexanoic acid (also known as phthaloyl amido-peroxy caproic acid), phthaloyl amido peroxy heptanoic acid, phthaloyl amido peroxy octanoic acid, phthaloyl amido peroxy nonanoic acid, phthaloyl amido peroxy decanoic acid and mixtures thereof.

[0029] Even more preferred peracids are any of having general formula:



wherein R is selected from C1-4 alkyl and n is an integer of from 1 to 5.

[0030] In a particularly preferred aspect of the present invention the peracid has the formula such that R is CH<sub>2</sub> and n is 5 i.e. phthaloyl amido-peroxy caproic acid or PAP.

[0031] The peracid is preferably used as a substantially water-insoluble solid or wetcake and is available from Ausimont under the trade name Euroco.

[0032] Other preferred peracids include N, N terephthaloyl di(6 amino percarboxy caproic acid) known as TPCAP.

[0033] In a particularly preferred embodiment of the present invention the peracid has mean average particle size

of less than 100 microns, more preferably less than 80 microns, even more preferably less than 60 microns. Most preferably PAP used herein has mean average particle size of between 20 and 50 microns.

[0034] The peracid is preferably present at a level of from 0.1% to 20%, more preferably from 1% to 10%, most preferably from 2% to 4%. In an alternative embodiment of the present invention the peracid may be present at a much higher level of for example 10% to 40%, more preferably from 15% to 30%, most preferably from 15% to 25%.

#### Co-agent

[0035] The compositions of the present invention comprise a co-agent. Co-agents as used herein refers to a compound selected from an organic acids, homo or copolymers of acidic monomers and mixtures thereof.

[0036] Suitable organic acids include those having molecular weight of less than 4000, more preferably less than 1000, most preferably less than 500.

[0037] In a particularly preferred embodiment the organic acid is selected from the group consisting of citric acid, maleic acid, oxalic acid, malonic acid, succinic acid, fumaric acid, tartaric acid, oxydiacetic acid, carboxymethyl oxy-succinic acid, carboxymethyl tartronic acid, ditartronic acid, oxydisuccinic acid, tartaric monosuccinic acid, tartaric disuccinic acid, guinaldic acid, picolinic acid, dipicolinic acid, C8-C22 saturated or unsaturated fatty acids and C8-C22 alkyl or alkenyl succinic acid, butane tri and tetracarboxylic acid, such as 1,2,3 butane tricarboxylic acid and 1,2,3,4 butane tetracarboxylic acid, and mixtures thereof. A preferred organic acid is citric acid.

[0038] Even more preferred organic acids are those having pKa1 equal to or below the pH of the composition into which they are to be formulated. Thus especially preferred organic acids include oxalic acid, malonic acid, succinic acid and mixtures thereof. These latter acids are di-carboxylic acids, having pKa1 of 1.23, 2.85 and 4.15, respectively. More preferably the organic acid is selected from oxalic acid or malonic acid.

[0039] Examples of suitable homopolymers include polymaleic acid, polyacrylic acid and polyglyoxalic acid. Examples of suitable copolymers include copolymers of maleic acid and acrylic acid. An alternative copolymer may contain more than two different monomeric units for example a polymer comprising maleic acid, acrylic acid and either acrylamide or acrylic esters.

[0040] The co-agent is present in the composition at a level of from 0.1% to 20%, more preferably from 1% to 10% and most preferably from 2% to 5%.

#### Optional ingredients

[0041] The compositions herein may further comprise a variety of other optional ingredients such as surfactants, suspending agent, chelating agents, radical scavengers, antioxidants, stabilisers, soil suspending polymer, polymeric soil release agents, pH control agents, dye transfer inhibitor, solvents, suds controlling agents, suds booster, brighteners, perfumes, pigments, dyes and the like.

#### Surfactant

[0042] The compositions of the present invention may comprise a surfactant as a preferred but optional ingredient. Where present a surfactant is selected from the group consisting of nonionic surfactants, anionic surfactants, zwitterionic surfactants, amphoteric surfactants, cationic surfactants and mixtures thereof.

[0043] Preferred anionic surfactants for use in the compositions herein include water-soluble salts or acids of the formula  $\text{ROSO}_3\text{M}$  wherein R preferably is a  $\text{C}_{10}\text{-C}_{24}$  hydrocarbyl, preferably an alkyl or hydroxyalkyl having a  $\text{C}_{10}\text{-C}_{20}$  alkyl component, more preferably a  $\text{C}_{12}\text{-C}_{18}$  alkyl or hydroxyalkyl, and M is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium), or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like). Typically, alkyl chains of  $\text{C}_{10}\text{-C}_{16}$ , more preferably  $\text{C}_{10}\text{-C}_{14}$  are preferred for lower wash temperatures (e.g. below approx.  $50^\circ\text{C}$ ) and  $\text{C}_{16}\text{-C}_{18}$  alkyl chains are preferred for higher wash temperatures (e.g. above approx.  $50^\circ\text{C}$ ).

[0044] Other suitable anionic surfactants for use herein are water-soluble salts or acids of the formula  $\text{RO(A)}_m\text{SO}_3\text{M}$  wherein R is an unsubstituted  $\text{C}_{10}\text{-C}_{24}$  alkyl or hydroxyalkyl group having a  $\text{C}_{10}\text{-C}_{24}$  alkyl component, preferably a  $\text{C}_{12}\text{-C}_{20}$  alkyl or hydroxyalkyl, more preferably  $\text{C}_{12}\text{-C}_{18}$  alkyl or hydroxyalkyl, A is an ethoxy or propoxy unit, m is greater than zero, typically between about 0.5 and about 6, more preferably between about 0.5 and about 3, and M is H or a cation which can be, for example, a metal cation (e.g., sodium, potassium, lithium, calcium, magnesium, etc.), ammonium or substituted-ammonium cation. Alkyl ethoxylated sulfates as well as alkyl propoxylated sulfates are contemplated herein. Specific examples of substituted ammonium cations include methyl-, dimethyl-, trimethyl-ammonium and quaternary ammonium cations, such as tetramethyl-ammonium, dimethyl piperdinium and cations derived from

alkanolamines such as ethylamine, diethylamine, triethylamine, mixtures thereof, and the like. Exemplary surfactants are C<sub>12</sub>-C<sub>18</sub> alkyl polyethoxylate (1.0) sulfate, C<sub>12</sub>-C<sub>18</sub>E(1.0)M, C<sub>12</sub>-C<sub>18</sub> alkyl polyethoxylate (2.25) sulfate, C<sub>12</sub>-C<sub>18</sub>E (2.25)M, C<sub>12</sub>-C<sub>18</sub> alkyl polyethoxylate (3.0) sulfate C<sub>12</sub>-C<sub>18</sub>E(3.0), and C<sub>12</sub>-C<sub>18</sub> alkyl polyethoxylate (4.0) sulfate C<sub>12</sub>-C<sub>18</sub>E(4.0)M, wherein M is conveniently selected from sodium and potassium.

**[0045]** Other particularly suitable anionic surfactants for use herein are alkyl sulphonates including water-soluble salts or acids of the formula RSO<sub>3</sub>M wherein R is a C<sub>6</sub>-C<sub>22</sub> linear or branched, saturated or unsaturated alkyl group, preferably a C<sub>12</sub>-C<sub>18</sub> alkyl group and more preferably a C<sub>14</sub>-C<sub>16</sub> alkyl group, and M is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium), or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like).

**[0046]** Suitable alkyl aryl sulphonates for use herein include water-soluble salts or acids of the formula RSO<sub>3</sub>M wherein R is an aryl, preferably a benzyl, substituted by a C<sub>6</sub>-C<sub>22</sub> linear or branched saturated or unsaturated alkyl group, preferably a C<sub>12</sub>-C<sub>18</sub> alkyl group and more preferably a C<sub>14</sub>-C<sub>16</sub> alkyl group, and M is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium, calcium, magnesium etc) or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like).

**[0047]** The alkylsulfonates and alkyl aryl sulphonates for use herein include primary and secondary alkylsulfonates and primary and secondary alkyl aryl sulphonates. By "secondary C<sub>6</sub>-C<sub>22</sub> alkyl or C<sub>6</sub>-C<sub>22</sub> alkyl aryl sulphonates", it is meant herein that in the formula as defined above, the SO<sub>3</sub>M or aryl-SO<sub>3</sub>M group is linked to a carbon atom of the alkyl chain being placed between two other carbons of the said alkyl chain (secondary carbon atom).

**[0048]** For example C<sub>14</sub>-C<sub>16</sub> alkyl sulphonate salt is commercially available under the name Hostapur® SAS from Hoechst and C<sub>8</sub>-alkylsulphonate sodium salt is commercially available under the name Witconate NAS 8® from Witco SA. An example of commercially available alkyl aryl sulphonate is Lauryl aryl sulphonate from Su.Ma. Particularly preferred alkyl aryl sulphonates are alkyl benzene sulphonates commercially available under trade name Nansa® available from Albright&Wilson.

**[0049]** Other anionic surfactants useful for deterative purposes can also be used herein. These can include salts (including, for example, sodium, potassium, ammonium, and substituted ammonium salts such as mono-, di- and tri-ethanolamine salts) of soap, C<sub>8</sub>-C<sub>24</sub> olefinsulfonates, sulfonated polycarboxylic acids prepared by sulfonation of the pyrolyzed product of alkaline earth metal citrates, e.g., as described in British patent specification No. 1,082,179, C<sub>8</sub>-C<sub>24</sub> alkylpolyglycoethersulfates (containing up to 10 moles of ethylene oxide); alkyl ester sulfonates such as C<sub>14</sub>-C<sub>16</sub> methyl ester sulfonates; acyl glycerol sulfonates, fatty oleyl glycerol sulfates, alkyl phenol ethylene oxide ether sulfates, paraffin sulfonates, alkyl phosphates, isethionates such as the acyl isethionates, N-acyl taurates, alkyl succinamates and sulfosuccinates, monoesters of sulfosuccinate (especially saturated and unsaturated C<sub>12</sub>-C<sub>18</sub> monoesters) di-

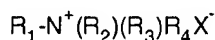
esters of sulfosuccinate (especially saturated and unsaturated C<sub>6</sub>-C<sub>14</sub> diesters), sulfates of alkylpolysaccharides such as the sulfates of alkylpolyglucoside (the nonionic nonsulfated compounds being described below), branched primary alkyl sulfates, alkyl polyethoxy carboxylates such as those of the formula RO(CH<sub>2</sub>CH<sub>2</sub>O)<sub>k</sub>CH<sub>2</sub>COO-M<sup>+</sup> wherein R is a C<sub>8</sub>-C<sub>22</sub> alkyl, k is an integer from 0 to 10, and M is a soluble salt-forming cation. Resin acids and hydrogenated resin acids are also suitable, such as rosin, hydrogenated rosin, and resin acids and hydrogenated resin acids present in or derived from tall oil. Further examples are given in "Surface Active Agents and Detergents" (Vol. I and II by Schwartz, Perry and Berch). A variety of such surfactants are also generally disclosed in U.S. Patent 3,929,678, issued December 30, 1975, to Laughlin, et al. at Column 23, line 58 through Column 29, line 23 (herein incorporated by reference).

**[0050]** Other particularly suitable anionic surfactants for use herein are alkyl carboxylates and alkyl alkoxycarboxylates having from 4 to 24 carbon atoms in the alkyl chain, preferably from 8 to 18 and more preferably from 8 to 16, wherein the alkoxy is propoxy and/or ethoxy and preferably is ethoxy at an alkoxylation degree of from 0.5 to 20, preferably from 5 to 15. Preferred alkylalkoxycarboxylate for use herein is sodium laureth 11 carboxylate (i.e., RO (C<sub>2</sub>H<sub>4</sub>O)<sub>10</sub>-CH<sub>2</sub>COONa, with R= C<sub>12</sub>-C<sub>14</sub>) commercially available under the name AkypoSoft® 100NV from Kao Chemical Gbmh.

**[0051]** Suitable amphoteric surfactants for use herein include amine oxides having the following formula R<sub>1</sub>R<sub>2</sub>R<sub>3</sub>NO wherein each of R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> is independently a saturated substituted or unsubstituted, linear or branched hydrocarbon chain of from 1 to 30 carbon atoms. Preferred amine oxide surfactants to be used according to the present invention are amine oxides having the following formula R<sub>1</sub>R<sub>2</sub>R<sub>3</sub>NO wherein R<sub>1</sub> is an hydrocarbon chain comprising from 1 to 30 carbon atoms, preferably from 6 to 20, more preferably from 8 to 16, most preferably from 8 to 12, and wherein R<sub>2</sub> and R<sub>3</sub> are independently substituted or unsubstituted, linear or branched hydrocarbon chains comprising from 1 to 4 carbon atoms, preferably from 1 to 3 carbon atoms, and more preferably are methyl groups. R<sub>1</sub> may be a saturated, substituted or unsubstituted linear or branched hydrocarbon chain. Suitable amine oxides for use herein are for instance natural blend C<sub>8</sub>-C<sub>10</sub> amine oxides as well as C<sub>12</sub>-C<sub>16</sub> amine oxides commercially available from Hoechst.

In a particularly preferred embodiment the surfactant where present is a C16 amine oxide.

**[0052]** Suitable zwitterionic surfactants for use herein contain both a cationic hydrophilic group, i.e., a quaternary ammonium group, and anionic hydrophilic group on the same molecule at a relatively wide range of pH's. The typical anionic hydrophilic groups are carboxylates and sulfonates, although other groups like sulfates, phosphonates, and the like can be used. A generic formula for the zwitterionic surfactants to be used herein is :



wherein  $R_1$  is a hydrophobic group;  $R_2$  is hydrogen,  $C_1$ - $C_6$  alkyl, hydroxy alkyl or other substituted  $C_1$ - $C_6$  alkyl group;  $R_3$  is  $C_1$ - $C_6$  alkyl, hydroxy alkyl or other substituted  $C_1$ - $C_6$  alkyl group which can also be joined to  $R_2$  to form ring structures with the N, or a  $C_1$ - $C_6$  carboxylic acid group or a  $C_1$ - $C_6$  sulfonate group;  $R_4$  is a moiety joining the cationic nitrogen atom to the hydrophilic group and is typically an alkylene, hydroxy alkylene, or polyalkoxy group containing from 1 to 10 carbon atoms; and X is the hydrophilic group which is a carboxylate or sulfonate group.

**[0053]** Preferred hydrophobic groups  $R_1$  are aliphatic or aromatic, saturated or unsaturated, substituted or unsubstituted hydrocarbon chains that can contain linking groups such as amido groups, ester groups. More preferred  $R_1$  is an alkyl group containing from 1 to 24 carbon atoms, preferably from 8 to 18, and more preferably from 10 to 16. These simple alkyl groups are preferred for cost and stability reasons. However, the hydrophobic group  $R_1$  can also be an amido radical of the formula  $R_a-C(O)-NH-(C(R_b)_2)_m$ , wherein  $R_a$  is an aliphatic or aromatic, saturated or unsaturated, substituted or unsubstituted hydrocarbon chain, preferably an alkyl group containing from 8 up to 20 carbon atoms, preferably up to 18, more preferably up to 16,  $R_b$  is selected from the group consisting of hydrogen and hydroxy groups, and m is from 1 to 4, preferably from 2 to 3, more preferably 3, with no more than one hydroxy group in any  $(C(R_b)_2)$  moiety.

**[0054]** Preferred  $R_2$  is hydrogen, or a  $C_1$ - $C_3$  alkyl and more preferably methyl. Preferred  $R_3$  is a  $C_1$ - $C_4$  carboxylic acid group or  $C_1$ - $C_4$  sulfonate group, or a  $C_1$ - $C_3$  alkyl and more preferably methyl. Preferred  $R_4$  is  $(CH_2)_n$  wherein n is an integer from 1 to 10, preferably from 1 to 6, more preferably is from 1 to 3.

**[0055]** Some common examples of betaine/sulphobetaine are described in U.S. Pat. Nos. 2,082,275, 2,702,279 and 2,255,082, incorporated herein by reference.

**[0056]** Examples of particularly suitable alkyldimethyl betaines include coconutdimethyl betaine, lauryl dimethyl betaine, decyl dimethyl betaine, 2-(N-decyl-N, N-dimethyl-ammonia)acetate, 2-(N-coco N, N-dimethylammonio) acetate, myristyl dimethyl betaine, palmityl dimethyl betaine, cetyl dimethyl betaine, stearyl dimethyl betaine. For example Coconut dimethyl betaine is commercially available from Seppic under the trade name of Amonyl 265®. Lauryl betaine is commercially available from Albright & Wilson under the trade name Empigen BB/L®.

**[0057]** Examples of amidobetaines include cocoamidoethylbetaine, cocoamidopropyl betaine or C10-C14 fatty acylamidopropylene(hydropropylene)sulfo betaine. For example C10-C14 fatty acylamidopropylene(hydropropylene)sulfo betaine is commercially available from Sherex Company under the trade name "Varion CAS® sulfobetaine".

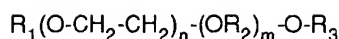
**[0058]** A further example of betaine is Lauryl-immino-dipropionate commercially available from Rhone-Poulenc under the trade name Mirataine H2C-HA®.

**[0059]** Suitable cationic surfactants for use herein include derivatives of quaternary ammonium, phosphonium, imidazolium and sulfonium compounds. Preferred cationic surfactants for use herein are quaternary ammonium compounds wherein one or two of the hydrocarbon groups linked to nitrogen are a saturated, linear or branched alkyl group of 6 to 30 carbon atoms, preferably of 10 to 25 carbon atoms, and more preferably of 12 to 20 carbon atoms, and wherein the other hydrocarbon groups (i.e. three when one hydrocarbon group is a long chain hydrocarbon group as mentioned hereinbefore or two when two hydrocarbon groups are long chain hydrocarbon groups as mentioned hereinbefore) linked to the nitrogen are independently substituted or unsubstituted, linear or branched, alkyl chain of from 1 to 4 carbon atoms, preferably of from 1 to 3 carbon atoms, and more preferably are methyl groups. Preferred quaternary ammonium compounds suitable for use herein are non-chloride/non halogen quaternary ammonium compounds. The counterion used in said quaternary ammonium compounds are compatible with any persulfate salt and are selected from the group of methyl sulfate, or methylsulfonate, and the like.

**[0060]** Particularly preferred for use in the compositions of the present invention are trimethyl quaternary ammonium compounds like myristyl trimethylsulfate, cetyl trimethylsulfate and/or tallow trimethylsulfate. Such trimethyl quaternary ammonium compounds are commercially available from Hoechst, or from Albright & Wilson under the trade name EMPIGEN CM®.

**[0061]** Amongst the nonionic surfactants, alkoxyated nonionic surfactants and especially ethoxyated nonionic surfactants are suitable for use herein. Particularly preferred nonionic surfactants for use herein are the capped alkoxyated nonionic surfactants as they have improved stability to the persulfate salts.

**[0062]** Suitable capped alkoxyated nonionic surfactants for use herein are according to the formula:



wherein  $R_1$  is a  $C_8$ - $C_{24}$  linear or branched alkyl or alkenyl group, aryl group, alkaryl group, preferably  $R_1$  is a  $C_8$ - $C_{18}$  alkyl or alkenyl group, more preferably a  $C_{10}$ - $C_{15}$  alkyl or alkenyl group, even more preferably a  $C_{10}$ - $C_{15}$  alkyl group;

wherein  $R_2$  is a  $C_1$ - $C_{10}$  linear or branched alkyl group, preferably a  $C_2$ - $C_{10}$  linear or branched alkyl group ;

wherein  $R_3$  is a  $C_1$ - $C_{10}$  alkyl or alkenyl group, preferably a  $C_1$ - $C_5$  alkyl group, more preferably methyl;

and wherein  $n$  and  $m$  are integers independently ranging in the range of from 1 to 20, preferably from 1 to 10, more preferably from 1 to 5; or mixtures thereof.

[0063] These surfactants are commercially available from BASF under the trade name Plurafac®, from HOECHST under the trade name Genapol® or from ICI under the trade name Symperonic®. Preferred capped nonionic alkoxyated surfactants of the above formula are those commercially available under the tradename Genapol® L 2.5 NR from Hoechst, and Plurafac® from BASF.

[0064] Particularly preferred surfactants are those selected from the group consisting of alkyl sulphate, alkyl ethoxy sulphate, alkyl sulphonate, alkyl benzene sulphonate, alkyl carboxylate, alkyl ethoxy carboxylate and mixtures thereof.

[0065] Where present the compositions according to the present invention comprise surfactant at a level of 5% or less, preferably from 0.01% to 5%, more preferably at a level of from 0.1% to 4 % and most preferably from 0.2% to 3% by weight of the composition.

[0066] According to another aspect of the present invention there is provided a product comprising two compositions, one of which comprises surfactant. This surfactant composition may comprise any of the surfactants as discussed previously. The surfactant or combination of surfactants of this composition makes up for 10-40% of this surfactant-containing composition.

#### Suspending agent

[0067] The composition of the present invention may preferably comprise a suspending agent. A suspending agent is an ingredient which is specifically added to the composition of the present invention to suspend a solid particulate ingredient of the composition. With regard to the present invention, a suspending agent is particularly useful for suspending the PAP.

[0068] Suitable suspending agents are those known in the art. Examples of suspending agents include gum-type polymers (e.g. xanthan gum), polyvinyl alcohol and derivatives thereof, cellulose and derivatives thereof and polycarboxylate polymers.

[0069] In a particularly preferred embodiment of the present invention, the suspending agent is selected from a gum-type polymer or a polycarboxylate polymer.

[0070] The gum-type polymer may be selected from the group consisting of polysaccharide hydrocolloids, xanthan gum, guar gum, succinoglucon gum, Cellulose, derivatives of any of the above and mixtures thereof. In a preferred aspect of the present invention the gum-type polymer is a xanthan gum or derivative thereof.

[0071] The gum-type polymer is preferably present at a level of from 0.01% to 10%, more preferably from 0.1 % to 3%.

[0072] The polycarboxylate polymer can be a homo or copolymer of monomer units selected from acrylic acid, methacrylic acid, maleic acid, malic acid, maleic anhydride. Preferred polycarboxylate polymers are Carbopol from BF Goodrich. Suitable polymers have molecular weight in the range of from 10 000 to 10 000 000, more preferably 100 000 to 10 000 000.

[0073] The cross-linked polycarboxylate polymer is preferably present at a level of from 0.01% to 2%, more preferably from 0.01% to 1%, most preferably from 0.1% to 0.8%.

[0074] The ratio of gum-type polymer to cross-linked polycarboxylate polymer is from 100:1 to 1:100, more preferably from 10:1 to 1:10.

#### Chelating agents

[0075] The compositions of the present invention may comprise a chelating agent as a preferred optional ingredient. Suitable chelating agents may be any of those known to those skilled in the art such as the ones selected from the group comprising phosphonate chelating agents, amino carboxylate chelating agents, other carboxylate chelating agents, polyfunctionally-substituted aromatic chelating agents, ethylenediamine N,N'- disuccinic acids, or mixtures



thereof.

**[0076]** The presence of chelating agents contribute to further enhance the chemical stability of the compositions. A chelating agent may be also desired in the compositions of the present invention as it allows to increase the ionic strength of the compositions herein and thus their stain removal and bleaching performance on various surfaces.

**[0077]** Suitable phosphonate chelating agents for use herein may include alkali metal ethane 1-hydroxy diphosphonates (HEDP), alkylene poly (alkylene phosphonate), as well as amino phosphonate compounds, including amino aminotri(methylene phosphonic acid) (ATMP), nitrilo trimethylene phosphonates (NTP), ethylene diamine tetra methylene phosphonates, and diethylene triamine penta methylene phosphonates (DTPMP). The phosphonate compounds may be present either in their acid form or as salts of different cations on some or all of their acid functionalities. Preferred phosphonate chelating agents to be used herein are diethylene triamine penta methylene phosphonate (DTPMP) and ethane 1-hydroxy diphosphonate (HEDP). Such phosphonate chelating agents are commercially available from Monsanto under the trade name DEQUEST®.

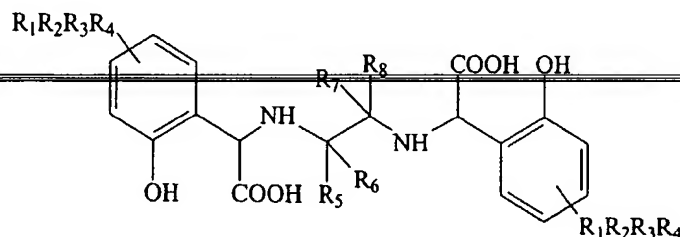
**[0078]** Polyfunctionally-substituted aromatic chelating agents may also be useful in the compositions herein. See U. S. patent 3,812,044, issued May 21, 1974, to Connor et al. Preferred compounds of this type in acid form are dihydroxydisulfobenzenes such as 1,2-dihydroxy-3,5-disulfobenzene.

**[0079]** A preferred biodegradable chelating agent for use herein is ethylene diamine N,N'-disuccinic acid, or alkali metal, or alkaline earth, ammonium or substitutes ammonium salts thereof or mixtures thereof. Ethylenediamine N,N'-disuccinic acids, especially the (S,S) isomer have been extensively described in US patent 4, 704, 233, November 3, 1987, to Hartman and Perkins. Ethylenediamine N,N'-disuccinic acids is, for instance, commercially available under the tradename ssEDDS® from Palmer Research Laboratories.

**[0080]** Suitable amino carboxylates to be used herein include ethylene diamine tetra acetates, diethylene triamine pentaacetates, diethylene triamine pentaacetate (DTPA), N-hydroxyethylethylenediamine triacetates, nitrilotriacetates, ethylenediamine tetrapropionates, triethylenetetraaminehexa-acetates, ethanoldiglycines, propylene diamine tetracetic acid (PDTA) and methyl glycine di-acetic acid (MGDA), both in their acid form, or in their alkali metal, ammonium, and substituted ammonium salt forms. Particularly suitable amino carboxylates to be used herein are diethylene triamine penta acetic acid, propylene diamine tetracetic acid (PDTA) which is, for instance, commercially available from BASF under the trade name Trilon FS® and methyl glycine di-acetic acid (MGDA).

**[0081]** Further carboxylate chelating agents to be used herein include salicylic acid, aspartic acid, glutamic acid, glycine, malonic acid or mixtures thereof.

**[0082]** Another chelating agent for use herein is of the formula:



wherein  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of -H, alkyl, alkoxy, aryl, aryloxy, -Cl, -Br, -NO<sub>2</sub>, -C(O)R', and -SO<sub>2</sub>R"; wherein R' is selected from the group consisting of -H, -OH, alkyl, alkoxy, aryl, and aryloxy; R" is selected from the group consisting of alkyl, alkoxy, aryl, and aryloxy; and  $R_5$ ,  $R_6$ ,  $R_7$ , and  $R_8$  are independently selected from the group consisting of -H and alkyl.

**[0083]** Particularly preferred chelating agents to be used herein are amino aminotri(methylene phosphonic acid), diethylene-triamino-pentaacetic acid, diethylene triamine penta methylene phosphonate, 1-hydroxy ethane diphosphonate, ethylenediamine N, N'-disuccinic acid, and mixtures thereof.

**[0084]** Typically, the compositions according to the present invention comprise up to 5% by weight of the total composition of a chelating agent, or mixtures thereof, preferably from 0.01% to 1.5% by weight and more preferably from 0.01% to 0.5%.

#### Radical scavengers

**[0085]** The compositions of the present invention may comprise a radical scavenger or a mixture thereof.

**[0086]** Suitable radical scavengers for use herein include the well-known substituted mono and dihydroxy benzenes and their analogs, alkyl and aryl carboxylates and mixtures thereof. Preferred such radical scavengers for use herein include di-tert-butyl hydroxy toluene (BHT), hydroquinone, di-tert-butyl hydroquinone, mono-tert-butyl hydroquinone,

tert-butyl-hydroxy anisole, benzoic acid, toluic acid, catechol, t-butyl catechol, benzylamine, 1,1,3-tris(2-methyl-4-hydroxy-5-t-butylphenyl) butane, n-propyl-gallate or mixtures thereof and highly preferred is di-tert-butyl hydroxy toluene. Such radical scavengers like N-propyl-gallate may be commercially available from Nipa Laboratories under the trade name Nipanox S1 ®.

5 [0087] Radical scavengers when used, are typically present herein in amounts up to 10% by weight of the total composition and preferably from 0.001% to 0.5% by weight.

[0088] The presence of radical scavengers may contribute to the chemical stability of the bleaching compositions of the present invention as well as to the safety profile of the compositions of the present invention.

#### 10 Suds controlling agents

[0089] The compositions according to the present invention may further comprise a suds controlling agent such as 2-alkyl alkanol, or mixtures thereof, as a preferred optional ingredient. Particularly suitable to be used in the present invention are the 2-alkyl alkanols having an alkyl chain comprising from 6 to 16 carbon atoms, preferably from 8 to 12 and a terminal hydroxy group, said alkyl chain being substituted in the  $\alpha$  position by an alkyl chain comprising from 1 to 10 carbon atoms, preferably from 2 to 8 and more preferably 3 to 6. Such suitable compounds are commercially available, for instance, in the Isofol® series such as Isofol® 12 (2-butyl octanol) or Isofol® 16 (2-hexyl decanol).

[0090] Other suds controlling agents may include alkali metal (e.g., sodium or potassium) fatty acids, or soaps thereof, containing from about 8 to about 24, preferably from about 10 to about 20 carbon atoms.

20 [0091] The fatty acids including those used in making the soaps can be obtained from natural sources such as, for instance, plant or animal-derived glycerides (e.g., palm oil, coconut oil, babassu oil, soybean oil, castor oil, tallow, whale oil, fish oil, tallow, grease, lard and mixtures thereof). The fatty acids can also be synthetically prepared (e.g., by oxidation of petroleum stocks or by the Fischer-Tropsch process).

[0092] Alkali metal soaps can be made by direct saponification of fats and oils or by the neutralization of the free fatty acids which are prepared in a separate manufacturing process. Particularly useful are the sodium and potassium salts of the mixtures of fatty acids derived from coconut oil and tallow, i.e., sodium and potassium tallow and coconut soaps.

30 [0093] The term "tallow" is used herein in connection with fatty acid mixtures which typically have an approximate carbon chain length distribution of 2.5% C14, 29% C16, 23% C18, 2% palmitoleic, 41.5% oleic and 3% linoleic (the first three fatty acids listed are saturated). Other mixtures with similar distribution, such as the fatty acids derived from various animal tallow and lard, are also included within the term tallow. The tallow can also be hardened (i.e., hydrogenated) to convert part or all of the unsaturated fatty acid moieties to saturated fatty acid moieties.

[0094] When the term "coconut" is used herein it refers to fatty acid mixtures which typically have an approximate carbon chain length distribution of about 8% C8, 7% C10, 48% C12, 17% C14, 9% C16, 2% C18, 7% oleic, and 2% linoleic (the first six fatty acids listed being saturated). Other sources having similar carbon chain length distribution such as palm kernel oil and babassu oil are included with the term coconut oil.

40 [0095] Other suitable suds controlling agents are exemplified by silicones, and silica-silicone mixtures. Silicones can be generally represented by alkylated polysiloxane materials while silica is normally used in finely divided forms exemplified by silica aerogels and xerogels and hydrophobic silicas of various types. These materials can be incorporated as particulates in which the suds controlling agent is advantageously releasably incorporated in a water-soluble or water-dispersible, substantially non-surface-active detergent impermeable carrier. Alternatively the suds controlling agent can be dissolved or dispersed in a liquid carrier and applied by spraying on to one or more of the other components.

45 [0096] A preferred silicone suds controlling agent is disclosed in Bartollota et al. U.S. Patent 3 933 672. Other particularly useful suds controlling agents are the self-emulsifying silicone suds controlling agents, described in German Patent Application DTOS 2 646 126 published April 28, 1977. An example of such a compound is DC-544, commercially available from Dow Corning, which is a siloxane-glycol copolymer.

[0097] Especially preferred silicone suds controlling agents are described in Copending European Patent application N°92201649.8. Said compositions can comprise a silicone/silica mixture in combination with fumed nonporous silica such as Aerosil®.

[0098] Especially preferred suds controlling agent are the suds controlling agent system comprising a mixture of silicone oils and the 2-alkyl-alkanols.

[0099] Typically, the compositions herein may comprise up to 4% by weight of the total composition of a suds controlling agent, or mixtures thereof, preferably from 0.1 % to 1.5% and most preferably from 0.1% to 0.8%.

#### 55 Stabilisers

[0100] The compositions of the present invention may further comprise up to 10%, preferably from 2% to 4% by

weight of the total composition of an alcohol according to the formula  $\text{HO} - \text{CR}'\text{R}'' - \text{OH}$ , wherein  $\text{R}'$  and  $\text{R}''$  are independently H or a C2-C10 hydrocarbon chain and/or cycle. Preferred alcohol according to that formula is propanediol. Indeed, we have observed that these alcohols in general and propanediol in particular also improve the chemical stability of the compositions.

**[0101]** Other stabilizers like inorganic stabilizers may be used herein. Examples of inorganic stabilizers include sodium stannate and various alkali metal phosphates such as the well-known sodium tripolyphosphates, sodium pyrophosphate and sodium orthophosphate.

#### Soil suspending polymer

**[0102]** The compositions herein may also comprise other polymeric soil release agents known to those skilled in the art. Such polymeric soil release agents are characterised by having both hydrophilic segments, to hydrophilize the surface of hydrophobic fibres, such as polyester and nylon, and hydrophobic segments, to deposit upon hydrophobic fibres and remain adhered thereto through completion of washing and rinsing cycles and, thus, serve as an anchor for the hydrophilic segments. This can enable stains occurring subsequent to treatment with the soil release agent to be more easily cleaned in later washing procedures.

**[0103]** The polymeric soil release agents useful herein especially include those soil release agents having: (a) one or more nonionic hydrophile components consisting essentially of (i) polyoxyethylene segments with a degree of polymerization of at least 2, or (ii) oxypropylene or polyoxypropylene segments with a degree of polymerization of from 2 to 10, wherein said hydrophile segment does not encompass any oxypropylene unit unless it is bonded to adjacent moieties at each end by ether linkages, or (iii) a mixture of oxyalkylene units comprising oxyethylene and from 1 to about 30 oxypropylene units wherein said mixture contains a sufficient amount of oxyethylene units such that the hydrophile component has hydrophilicity great enough to increase the hydrophilicity of conventional polyester synthetic fiber surfaces upon deposit of the soil release agent on such surface, said hydrophile segments preferably comprising at least about 25% oxyethylene units and more preferably, especially for such components having about 20 to 30 oxypropylene units, at least about 50% oxyethylene units; or (b) one or more hydrophobe components comprising (i) C<sub>3</sub> oxyalkylene terephthalate segments, wherein, if said hydrophobe components also comprise oxyethylene terephthalate, the ratio of oxyethylene terephthalate:C<sub>3</sub> oxyalkylene terephthalate units is about 2:1 or lower, (ii) C<sub>4</sub>-C<sub>6</sub> alkylene or oxy C<sub>4</sub>-C<sub>6</sub> alkylene segments, or mixtures therein, (iii) poly (vinyl ester) segments, preferably polyvinyl acetate, having a degree of polymerization of at least 2, or (iv) C<sub>1</sub>-C<sub>4</sub> alkyl ether or C<sub>4</sub> hydroxyalkyl ether substituents, or mixtures therein, wherein said substituents are present in the form of C<sub>1</sub>-C<sub>4</sub> alkyl ether or C<sub>4</sub> hydroxyalkyl ether cellulose derivatives, or mixtures therein, and such cellulose derivatives are amphiphilic, whereby they have a sufficient level of C<sub>1</sub>-C<sub>4</sub> alkyl ether and/or C<sub>4</sub> hydroxyalkyl ether units to deposit upon conventional polyester synthetic fiber surfaces and retain a sufficient level of hydroxyls, once adhered to such conventional synthetic fiber surface, to increase fiber surface hydrophilicity, or a combination of (a) and (b).

**[0104]** Typically, the polyoxyethylene segments of (a)(i) will have a degree of polymerization of from about 1 to about 200, although higher levels can be used, preferably from 3 to about 150, more preferably from 6 to about 100. Suitable oxy C<sub>4</sub>-C<sub>6</sub> alkylene hydrophobe segments include, but are not limited to, end-caps of polymeric soil release agents such as  $\text{MO}_3\text{S}(\text{CH}_2)_n\text{OCH}_2\text{CH}_2\text{O}-$ , where M is sodium and n is an integer from 4-6, as disclosed in U.S. Patent 4,721,580, issued January 26, 1988 to Gosselink.

**[0105]** Polymeric soil release agents useful in the present invention also include cellulosic derivatives such as hydroxyether cellulosic polymers, co-polymeric blocks of ethylene terephthalate or propylene terephthalate with polyethylene oxide or polypropylene oxide terephthalate, and the like. Such agents are commercially available and include hydroxyethers of cellulose such as METHOCEL (Dow). Cellulosic soil release agents for use herein also include those selected from the group consisting of C<sub>1</sub>-C<sub>4</sub> alkyl and C<sub>4</sub> hydroxyalkyl cellulose; see U.S. Patent 4,000,093, issued December 28, 1976 to Nicol, et al.

**[0106]** Soil release agents characterised by poly(vinyl ester) hydrophobe segments include graft co-polymers of poly (vinyl ester), e.g., C<sub>1</sub>-C<sub>6</sub> vinyl esters, preferably poly(vinyl acetate) grafted onto polyalkylene oxide backbones, such as polyethylene oxide backbones. See European Patent Application 0 219 048, published April 22, 1987 by Kud, et al. Commercially available soil release agents of this kind include the SOKALAN type of material, e.g., SOKALAN HP-22, available from BASF (West Germany).

**[0107]** One type of preferred soil release agent is a co-polymer having random blocks of ethylene terephthalate and polyethylene oxide (PEO) terephthalate. The molecular weight of this polymeric soil release agent is in the range of from about 25,000 to about 55,000. See U.S. Patent 3,959,230 to Hays, issued May 25, 1976 and U.S. Patent 3,893,929 to Basadur issued July 8, 1975.

**[0108]** Another preferred polymeric soil release agent is a polyester with repeat units of ethylene terephthalate units which contains 10-15% by weight of ethylene terephthalate units together with 90-80% by weight of polyoxyethylene terephthalate units, derived from a polyoxyethylene glycol of average molecular weight 300-5,000. Examples of this

polymer include the commercially available material ZELCON 5126 (from Dupont) and MILEASE T (from ICI). See also U.S. Patent 4,702,857, issued October 27, 1987 to Gosselink.

[0109] Another preferred polymeric soil release agent is a sulfonated product of a substantially linear ester oligomer comprised of an oligomeric ester backbone of terephthaloyl and oxyalkyleneoxy repeat units and terminal moieties covalently attached to the backbone. These soil release agents are fully described in U.S. Patent 4,968,451, issued November 6, 1990 to J.J. Scheibel and E.P. Gosselink. Other suitable polymeric soil release agents include the terephthalate polyesters of U.S. Patent 4,711,730, issued December 8, 1987 to Gosselink et al, the anionic end-capped oligomeric esters of U.S. Patent 4,721,580, issued January 26, 1988 to Gosselink, and the block polyester oligomeric compounds of U.S. Patent 4,702,857, issued October 27, 1987 to Gosselink.

[0110] Preferred polymeric soil release agents also include the soil release agents of U.S. Patent 4,877,896, issued October 31, 1989 to Maldonado et al, which discloses anionic, especially sulfoaroyl, end-capped terephthalate esters.

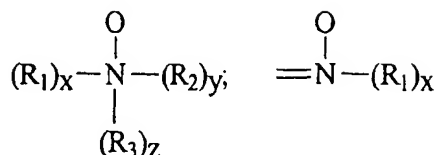
[0111] Still another preferred soil release agent is an oligomer with repeat units of terephthaloyl units, sulfoisoterephthaloyl units, oxyethyleneoxy and oxy-1,2-propylene units. The repeat units form the backbone of the oligomer and are preferably terminated with modified isethionate end-caps. A particularly preferred soil release agent of this type comprises about one sulfoisophthaloyl unit, 5 terephthaloyl units, oxyethyleneoxy and oxy-1,2-propyleneoxy units in a ratio of from about 1.7 to about 1.8, and two end-cap units of sodium 2-(2-hydroxyethoxy)-ethanesulfonate. Said soil release agent also comprises from about 0.5% to about 20%, by weight of the oligomer, of a crystalline-reducing stabilizer, preferably selected from the group consisting of xylene sulfonate, cumene sulfonate, toluene sulfonate, and mixtures thereof. See U.S. Pat. No. 5,415,807, issued May 16, 1995, to Gosselink et al.

[0112] If utilised, soil release agents will generally comprise from 0.01% to 10.0%, by weight, of the detergent compositions herein, typically from 0.1% to 5%, preferably from 0.2% to 3.0%.

#### Dye transfer inhibitor

[0113] The compositions of the present invention may also include one or more materials effective for inhibiting the transfer of dyes from one dyed surface to another during the cleaning process. Generally, such dye transfer inhibiting agents include polyvinyl pyrrolidone polymers, polyamine N-oxide polymers, co-polymers of N-vinylpyrrolidone and N-vinylimidazole, manganese phthalocyanine, peroxidases, and mixtures thereof. If used, these agents typically comprise from 0.01% to 10% by weight of the composition, preferably from 0.01% to 5%, and more preferably from 0.05% to 2%.

[0114] More specifically, the polyamine N-oxide polymers preferred for use herein contain units having the following structural formula:  $R-A_x-P$ ; wherein P is a polymerizable unit to which an N-O group can be attached or the N-O group can form part of the polymerizable unit or the N-O group can be attached to both units; A is one of the following structures:  $-NC(O)-$ ,  $-C(O)O-$ ,  $-S-$ ,  $-O-$ ,  $-N=$ ; x is 0 or 1; and R is aliphatic, ethoxylated aliphatics, aromatics, heterocyclic or alicyclic groups or any combination thereof to which the nitrogen of the N-O group can be attached or the N-O group is part of these groups. Preferred polyamine N-oxides are those wherein R is a heterocyclic group such as pyridine, pyrrole, imidazole, pyrrolidine, piperidine and derivatives thereof. The N-O group can be represented by the following general structures:



wherein  $R_1$ ,  $R_2$ ,  $R_3$  are aliphatic, aromatic, heterocyclic or alicyclic groups or combinations thereof; x, y and z are 0 or 1; and the nitrogen of the N-O group can be attached or form part of any of the aforementioned groups. The amine oxide unit of the polyamine N-oxides has a  $pK_a < 10$ , preferably  $pK_a < 7$ , more preferred  $pK_a < 6$ .

[0115] Any polymer backbone can be used as long as the amine oxide polymer formed is water-soluble and has dye transfer inhibiting properties. Examples of suitable polymeric backbones are polyvinyls, polyalkylenes, polyesters, polyethers, polyamide, polyimides, polyacrylates and mixtures thereof. These polymers include random or block co-polymers where one monomer type is an amine N-oxide and the other monomer type is an N-oxide. The amine N-oxide polymers typically have a ratio of amine to the amine N-oxide of 10:1 to 1:1,000,000. However, the number of amine oxide groups present in the polyamine oxide polymer can be varied by appropriate co-polymerization or by an appropriate degree of N-oxidation. The polyamine oxides can be obtained in almost any degree of polymerization. Typically, the average molecular weight is within the range of 500 to 1,000,000; more preferred 1,000 to 500,000; most preferred 5,000 to 100,000. This preferred class of materials can be referred to as "PVNO". The most preferred polyamine N-

oxide useful in the detergent compositions herein is poly(4-vinylpyridine-N-oxide) which as an average molecular weight of about 50,000 and an amine to amine N-oxide ratio of about 1:4.

[0116] Co-polymers of N-vinylpyrrolidone and N-vinylimidazole polymers (referred to as a class as "PVPVI") are also preferred for use herein. Preferably the PVPVI has an average molecular weight range from 5,000 to 1,000,000, more preferably from 5,000 to 200,000, and most preferably from 10,000 to 20,000. (The average molecular weight range is determined by light scattering as described in Barth, et al., *Chemical Analysis*, Vol 113. "Modern Methods of Polymer Characterization", the disclosures of which are incorporated herein by reference.) The PVPVI co-polymers typically have a molar ratio of N-vinylimidazole to N-vinylpyrrolidone from 1:1 to 0.2:1, more preferably from 0.8:1 to 0.3:1, most preferably from 0.6:1 to 0.4:1. These co-polymers can be either linear or branched.

[0117] The present invention compositions may also employ a polyvinylpyrrolidone ("PVP") having an average molecular weight of from 5,000 to 400,000, preferably from 5,000 to 200,000, and more preferably from 5,000 to 50,000. PVP's are known to persons skilled in the detergent field; see, for example, EP-A-262,897 and EP-A-256,696, incorporated herein by reference. Compositions containing PVP can also contain polyethylene glycol ("PEG") having an average molecular weight from 500 to 100,000, preferably from 1,000 to 10,000. Preferably, the ratio of PEG to PVP on a ppm basis delivered in wash solutions is from 2:1 to 50:1, and more preferably from 3:1 to 10:1.

#### Suds booster

[0118] If high sudsing is desired, suds boosters such as C<sub>10</sub>-C<sub>16</sub> alkanolamides can be incorporated into the compositions, typically at 1%-10% levels. The C<sub>10</sub>-C<sub>14</sub> monoethanol and diethanol amides illustrate a typical class of such suds boosters. Use of such suds boosters with high sudsing adjunct surfactants such as the amine oxides, betaines and sultaines noted above is also advantageous. If desired, soluble magnesium salts such as MgCl<sub>2</sub>, MgSO<sub>4</sub>, and the like, can be added at levels of, for example, 0.1%-2%, to provide additional suds and to enhance grease removal performance.

#### Brightener

[0119] Any optical brighteners, fluorescent whitening agents or other brightening or whitening agents known in the art can be incorporated in the instant compositions when they are designed for fabric treatment or laundering, at levels typically from about 0.05% to about 1.2%, by weight, of the detergent compositions herein. Commercial optical brighteners which may be useful in the present invention can be classified into subgroups, which include, but are not necessarily limited to, derivatives of stilbene, pyrazoline, coumarin, carboxylic acids, methinecyanines, dibenzothiophene-5,5-dioxide, azoles, 5- and 6-membered-ring heterocyclic brighteners, this list being illustrative and non-limiting. Examples of such brighteners are disclosed in "The Production and Application of Fluorescent Brightening Agents", M. Zahradnik, Published by John Wiley & Sons, New York (1982).

[0120] Specific examples of optical brighteners which are useful in the present compositions are those identified in U.S. Patent 4,790,856, issued to Wixon on December 13, 1988. These brighteners include the PHORWHITE series of brighteners from Verona. Other brighteners disclosed in this reference include: Tinopal UNPA, Tinopal CBS and Tinopal 5BM Tinopal PLC; available from Ciba-Geigy; Artic White CC and Artic White CWD, available from Hilton-Davis, located in Italy; the 2-(4-styryl-phenyl)-2H-naphthol[1,2-d]triazoles; 4,4'-bis-(1,2,3-triazol-2-yl)-stil-benes; 4,4'-bis(styryl)bisphenyls; and the aminocoumarins. Specific examples of these brighteners include 4-methyl-7-diethyl- amino coumarin; 1,2-bis-(benzimidazol-2-yl)ethylene; 2,5-bis(benzoxazol-2-yl)thiophene; 2-styryl-naph-[1,2-d]oxazole; and 2-(stilbene-4-yl)-2H-naphtho-[1,2-d]triazole. See also U.S. Patent 3,646,015, issued February 29, 1972, to Hamilton. Anionic brighteners are typically preferred herein.

#### pH control agent

[0121] A variety of suitable means can be used for adjusting the pH of the compositions, including organic or inorganic acids, alkanolamines and the like. It may be advantageous to use alkanolamines, in particular monoethanolamine, inasmuch as they have an additional effect of regulating the viscosity of the emulsion, without compromising on its physical stability.

#### Minor Ingredients

[0122] The composition described herein may also comprise minor ingredients such as pigment or dyes and perfumes.

Processes of treating surfaces

**[0123]** In the present invention, the liquid aqueous composition of the present invention can be used to clean surfaces. By "surfaces", it is meant herein any inanimate surface. These inanimate surfaces include, but are not limited to, hard-surfaces typically found in houses like kitchens, bathrooms, or in car interiors, e.g., tiles, walls, floors, chrome, glass, smooth vinyl, any plastic, plastified wood, table top, sinks, cooker tops, dishes, sanitary fittings such as sinks, showers, shower curtains, wash basins, WCs and especially include fabrics including clothes, curtains, drapes, bed linens, bath linens, table cloths, sleeping bags, tents, upholstered furniture and the like, and carpets. Inanimate surfaces also include household appliances including, but not limited to, refrigerators, freezers, washing machines, automatic dryers, ovens, microwave ovens, dishwashers and so on.

**[0124]** By "treating a surface", it is meant herein bleaching and/or disinfecting said surfaces as the compositions of the present invention comprise a bleaching system based on a peracid compound or a mixture thereof and optionally cleaning as said compositions may comprise a surfactant or any other conventional cleaning agents.

**[0125]** Thus, the present invention also encompasses a process of treating, especially bleaching a fabric, as the inanimate surface. In such a process a fabric is treated with a composition according to the present invention. Such treatment can be done either in a so-called "pretreatment mode", where a liquid bleaching composition, as defined herein, is applied neat onto said fabrics before the fabrics are rinsed, or washed with a conventional detergent, then rinsed, or in a "soaking mode" where a liquid bleaching composition, as defined herein, is first diluted in an aqueous bath and the fabrics are immersed and soaked in the bath, before preferably being rinsed, or in a "through the wash mode", where a liquid bleaching composition, as defined herein, is added in to a typical laundry detergent. It is preferred in both cases, that the fabrics be rinsed after they have been contacted with said composition and before said composition has completely dried off.

**[0126]** The processes of bleaching surfaces according to the present invention, especially fabrics, delivers effective whiteness performance as well as effective stain removal performance.

**[0127]** The compositions according to the present invention are preferably used to contact to fabrics in a liquid form. Indeed, by "in a liquid form", it is meant herein the liquid compositions according to the present invention per se in neat or in their diluted form.

**[0128]** The compositions according to the present invention are typically used in diluted form in a laundry operation. By "in diluted form", it is meant herein that the compositions for the bleaching of fabrics according to the present invention may be diluted by the user, preferably with water. Such dilution may occur for instance in hand laundry applications as well as by other means such as in a washing machine. Said compositions can be diluted up to 500 times, preferably from 5 to 200 times and more preferably from 10 to 80 times.

**[0129]** More specifically, a process of bleaching fabrics according to the present invention comprises the steps of first contacting said fabrics with a bleaching composition according to the present invention, in its diluted form, then ~~allowing said fabrics to remain in contact with said composition, for a period of time sufficient to bleach said fabrics,~~ typically 1 to 60 minutes, preferably 5 to 30 minutes, then rinsing said fabrics with water. If said fabrics are to be washed, i.e., with a conventional composition comprising at least one surfactant, said washing is preferably conducted together with the bleaching of said fabrics by contacting said fabrics at with a bleaching composition according to the present invention and said conventional surfactant-containing detergent composition at the same time. In an alternative process said washing may be conducted before or after said fabrics have been bleached with a bleaching composition according to the present invention. Accordingly, said process according to the present invention allows the user to bleach fabrics and optionally to wash fabrics with a detergent composition comprising at least one surfactant before the step of contacting said fabrics with said bleaching composition and/or in the step where said fabrics are contacted with said bleaching composition and/or after the step where said fabrics are contacted with said bleaching composition and before the rinsing step and/or after the rinsing step.

**[0130]** In a particularly preferred process of cleaning fabrics the bleaching composition and the surfactant containing conventional laundry detergent composition are applied to the fabric at the same time. Although it is essential that in this method the two compositions are not pre-mixed.

**[0131]** In another embodiment the present invention also encompasses a process of treating a hard-surface, as the inanimate surface. In such a process a composition, as defined herein, is contacted with the hard-surfaces to be treated. Thus, the present invention also encompasses a process of treating a hard-surface with a composition, as defined herein, wherein said process comprises the step of applying said composition to said hard-surface, preferably only soiled portions thereof, and optionally rinsing said hard-surface.

**[0132]** In the process of treating hard-surfaces according to the present invention the composition, as defined herein, may be applied to the surface to be treated in its neat form or in its diluted form typically up to 200 times their weight of water, preferably into 80 to 2 times their weight of water, and more preferably 60 to 2 times.

**[0133]** When used as hard surfaces bleaching/disinfecting compositions the compositions of the present invention are easy to rinse and provide good shine characteristics on the treated surfaces.

[0134] By "hard-surfaces", it is understood any hard-surfaces as mentioned herein before as well as dishes.

#### Packaging form of the liquid compositions:

[0135] Depending on the end-use envisioned, the compositions herein can be packaged in a variety of containers including conventional bottles, and bottles equipped with roll-on, sponge, brush or spray appliances.

[0136] In one particularly preferred embodiment of the present invention the composition is packaged in a two-compartment container, wherein the bleaching composition as described herein is packaged in one compartment and a second composition is packaged in the second compartment. In a particularly preferred aspect, the second composition is a conventional laundry detergent liquid composition, preferably comprising ingredients, particularly bleach-sensitive ingredients such as surfactants, enzymes and perfumes.

#### Examples

[0137] The invention is further illustrated by the following example which shows the stain removal benefits obtained by the combination of peracid bleach and citric acid.

[0138] The peracid used for the purposes of the present example is Phtaloylamido peroxyacaproic acid (PAP), obtained from the supplier (Ausimont) as a 70% wet cake. The wetcake PAP is primarily dispersed in distilled water under gentle agitation using a lab stirrer, to form a dispersion containing 2.7g of pure PAP in 100 mls water. A 1% solution of citric acid is then prepared and the pH is trimmed to 4.0 with sodium hydroxide. The PAP dispersion and the citric acid solution are then mixed together to obtain a dispersion of PAP in the citric acid solutions.

[0139] The liquid detergent used was a commercial product marketed by the Procter & Gamble Company under the tradename Liquid Tide and was produced in summer 1998. It was used at 92 mls /wash (99 grams) as recommended.

[0140] The compositions tested (in grams /wash) were:

- Liquid Tide (99g)
- Liquid Tide (99g) + 1 g citric acid
- Liquid Tide (99g) + 2.7 g PAP
- Liquid Tide (99 g) + PAP/citric acid dispersion containing 1 g citric acid + 2.7 g PAP

[0141] The PAP/citric acid dispersion was delivered into the washing machine at the same time as (but without premixing with) the liquid detergent. The tests were run using four identical Kenmore washing machines, using average US washing conditions, at a wash temperature of 32 C, using 64 liters water, of a hardness of 1 millimole/liter. The wash pH was measured and was similar for all four treatments (between 7.6 and 7.9). In each wash a piece of fabric with a total of 20 stains, was introduced, together with 1.5 kg of a white ballast load of mixed (cotton/polycotton) composition. Four cycles were run by scrambling the 4 products across the 4 washing machines; for each cycle new stains were used. At the end, the results of the four wash cycles were averaged.

[0142] The fabrics are visually graded by two expert graders using the 0-4 Panel Score Unit (psu) scale, where 0 means no difference and 4 corresponds to the highest difference. All psu grades are referred to the Liquid Tide control and positive grades mean a higher degree of stain removal vs Liquid Tide. The results were analyzed for statistical significance (at 95% confidence interval) and all grades that were significantly different vs. the Liquid Tide control are identified by the suffix "s". The results are shown here below:

	1	2	3	4
Treatment:	Tide	Tide + citric acid	Tide + PAP	Tide + citric acid + PAP
Stain removal Performance (psu)				
Stains: Wine	0.0	+ 0.1	+0.5	+ 1.1 s
Coffee	0.0	+ 0.1	+0.6	+ 0.9 s
Blueberry	0.0	+ 0.2	+0.7	+ 1.3 s
Blackcurrant	0.0	+ 0.5	+0.8	+ 1.6 s
Artificial menstrual fluid	0.0	-0.2	+ 0.7 s	+ 0.9 s
US Clay (on Polyester)	0.0	+ 0.4	-0.3	+ 1.2 s
Average on all (20) stains	0.0	0.0	+0.3	+ 0.7 s



[0143] The test shows that there is a synergy between PAP and citric acid, as the performance improvements of the PAP /citric acid combination is higher than the sum of the benefits obtained with citric acid alone and PAP alone.

[0144] The present formulations are provided to further exemplify the present invention, but are not intended to be limiting in any way.

Ingredient	Ex. 1	Ex. 2	Ex. 3	Ex. 4
	% w/w	% w/w	% w/w	% w/w
PAP	3	3	5	5
Xanthan Gum	0.3	0.3	0.3	0.3
Carbopol	0.2	0.2	0.2	0.2
C16AO	1	-	1	-
C7/9 sulphate	3	-	3	-
C12/14E3S	-	3	-	3
Tinopal SOP	0.02	0.02	0.02	0.02
Dye/pigment	0.01	0.01	0.01	0.01
Perfume	0.2	0.2	0.2	0.2
Oxalic acid	0.1	-	0.1	-
Malonic acid	-	0.1	0.1	0.1

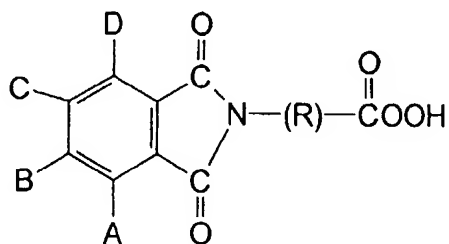
## Claims

1. A liquid bleaching composition comprising a pre-formed particulate peroxy carboxylic acid containing at least one aromatic group and a co-agent selected from the group consisting of organic acid, homo and copolymers of acidic monomers and mixtures thereof.
2. A liquid bleaching composition according to claim 1 wherein the peroxy carboxylic acid has general formula:



wherein Q is an aromatic group, A is  $P-(R)_n-C(O)OOH$  wherein R is an alkyl group, n is from 1 to 8, preferably 1 to 5, P is either CO-NH or C(O)NC(O) and m is either 0 or 1.

3. A liquid bleaching composition according to any preceding claim wherein the peroxy carboxylic acid has general formula:



where R is C1-20 and where A, B, C and D are independently either hydrogen or substituent groups individually selected from the group consisting of alkyl, hydroxyl, nitro, halogen, amine, ammonium, cyanide, carboxylic, sulphate, sulphonate, aldehydes or mixtures thereof.

4. A liquid bleaching composition according to any preceding claim wherein the peroxy carboxylic acid is selected



from phthaloyl amido-peroxy caproic acid, N,N terephthaloyl di(6 amino percarboxy caproic acid) and mixtures thereof.

- 5 5. A liquid bleaching composition according to any preceding claim wherein the organic acid is selected from the group consisting of citric acid, maleic acid, oxalic acid, malonic acid, succinic acid, fumaric acid, tartaric acid, oxydiacetic acid, carboxymethyl oxysuccinic acid, carboxymethyl tartronic acid, ditartronic acid, oxydisuccinic acid, tartaric monosuccinic acid, tartaric disuccinic acid, guinaldic acid, picolinic acid, dipicolinic acid, C8-C22 saturated or unsaturated fatty acids, C8-C22 alkyl or alkenyl succinic acid, butane tri and tetra carboxylic acid and mixtures thereof, preferably the organic acid is selected from citric acid, oxalic acid, malonic acid, succinic acid and mixtures thereof.
- 10 6. A liquid bleaching composition according to any preceding claim wherein the organic acid is selected such that the pKa1 of the acid is equal to or below the pH of the composition.
- 15 7. A liquid bleaching composition according to any of the preceding claims wherein the polymer a homo and/or co-polymers of monomer units selected from the group consisting of acrylic acid, maleic acid and polyglyoxalic acid.
8. A liquid bleaching composition according to any preceding claim wherein the co-agent is present at a level of from 0.1% to 20% by weight of the composition.
- 20 9. A liquid bleaching composition according to any preceding claim wherein the pH of the composition is in the range of from 1 to 6.
- 25 10. A liquid bleaching composition additionally comprising a surfactant selected from the group consisting of anionic, nonionic, amphoteric, zwitterionic, cationic surfactants and mixtures thereof at a level of 5% or less.
11. A liquid bleaching composition according to claim 10 wherein the surfactant is an amine oxide, more preferably a C16 amine oxide.
- 30 12. A liquid bleaching product comprising a first composition comprising a peroxy carboxylic acid containing at least one aromatic group and a second composition comprising a surfactant wherein the product is packaged such that first and second compositions are separated from each other.
- 35 13. A liquid bleaching product according to claim 12 wherein the first composition has a pH of from 1 to 6 and the second composition has a pH of 7 or more.
14. A process for cleaning fabrics by applying a liquid composition according to any of the preceding claims and optionally rinsing the fabrics.
- 40 15. A process according to claim 14, wherein a liquid bleaching composition according to any of claims 1 to 13 and a surfactant-containing composition are applied to a fabric without premixing the liquid bleaching or surfactant-containing compositions.
- 45 16. A process according to claim 15 wherein the liquid bleaching composition is applied to the fabric before, during or after the surfactant-containing composition is applied to the fabric.

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